

# Enrichment of Relevant Oxidative Degradation Products in Pharmaceuticals with Targeted Chemoselective Oxidation



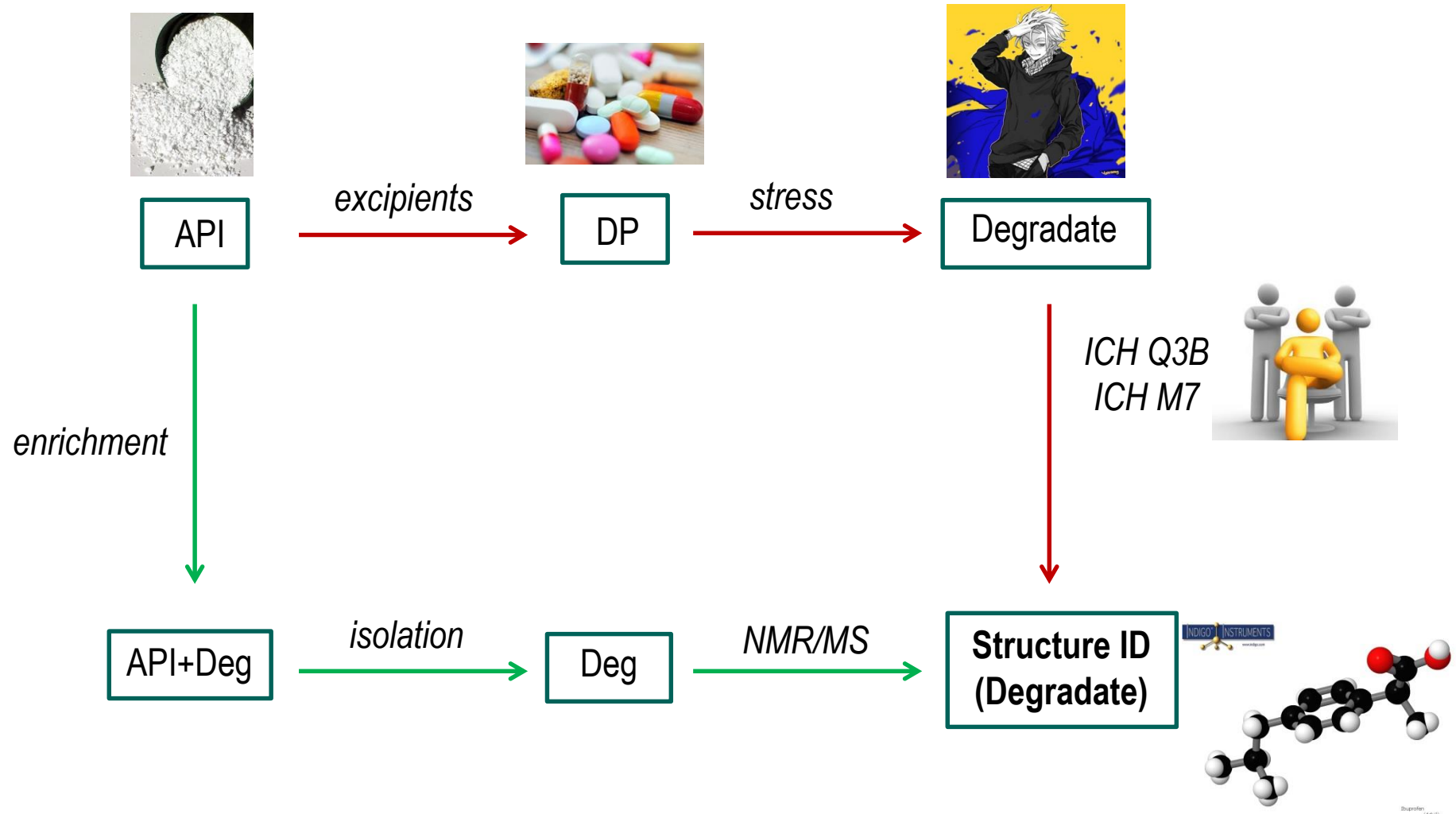
**MSD**

**INVENTING FOR LIFE**

October, 2019

Kausik Nanda

Merck Sharp & Dohme Corp.



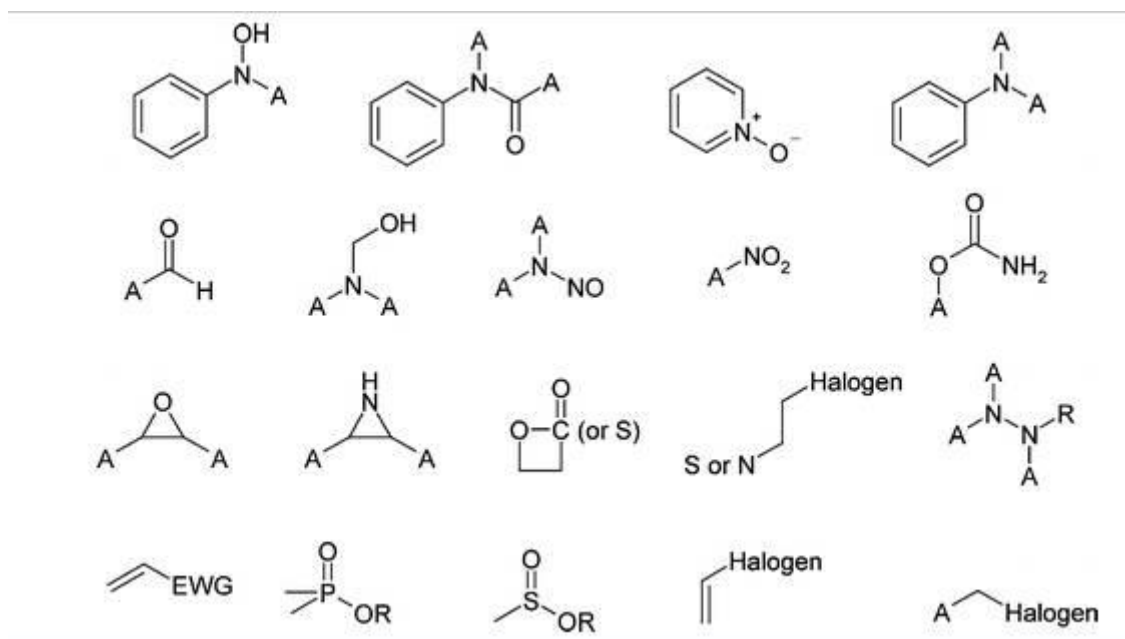
# Reporting, ID and Qualification Triggers

Maximum Daily Dose	Reporting Threshold	Identification Threshold	Qualification Threshold
<b>Drug Substance Impurities (ICH Q3A)</b>			
≤2 g/day	0.05%	0.10% or 1.0 mg per day*	0.15% or 1.0 mg/day*
>2 g/day	0.03%	0.05%	0.05%
<b>Drug Product Degradation Products (ICH Q3b)</b>			
≤1 g	0.10%		
>1 g	0.05%		
<1 mg		1.0% or 5 µg/day*	
1 mg–10 mg		0.5% or 20 µg/day*	
>10 mg–2 g		0.2% or 2 mg/day*	
>2 g		0.10%	
<10 mg			1.0% or 50 µg/day*
10 mg–100 mg			0.5% or 200 µg/day*
>100 mg–2 g			0.2% or 3 mg/day*
>2 g			0.15%

\*Whichever is lower

# ICH M7 and Mutagenic Impurity (MI) assessment

ICH M7 describes expectations for drug products on how degradation products should undergo mutagenicity assessment.



Example of an intercalation complex with DNA

2

Out-of-scope: Oncology<sup>^</sup>, oligonucleotides\*, peptides\*, APIs derived from a fermentation process, biologics.

<sup>^</sup>materials dosed in healthy volunteers are in scope

\* (e.g. linkers conjugating to PEG), or delivery system (e.g. lipid nanoparticle), **will be in scope for ICH M7**, if the specific chemical modifications have not previously been assessed

# ICH M7 and Mutagenic Impurities

Impurities classification (ICH M7)

Class	Definition
1	Known Mutagenic carcinogens
2	Known mutagen with unknown carcinogenic potential
3	Alerting structure, unrelated to DS; no mutagenicity data
4	Alerting structure, same alert in DS which have been tested and are non-mutagenic
5	No structural alerts

Acceptable intakes for an individual impurity

Duration of treatment	≤ 1 month	>1 – 12 months	>1 – 10 years	>10 years
Daily intake (µg/day)	120	20	10	1.5

# Common degradation pathways

1. **Hydrolytic**
2. **Oxidative**
3. **Thermal**
4. **Photolytic**

# Oxidative Degradation

## One Electron (radical intermediate)

Reactive species:  $\dot{O}R$   $\dot{O}OR$

*Forced stress with AIBN*



API

## Two-electrons

Reactive species: ROOH

*Forced stress with H<sub>2</sub>O<sub>2</sub>*

*enrichment*



## Transition Metal

Most abundant metal ion: Fe<sup>3+</sup>

API+Deg

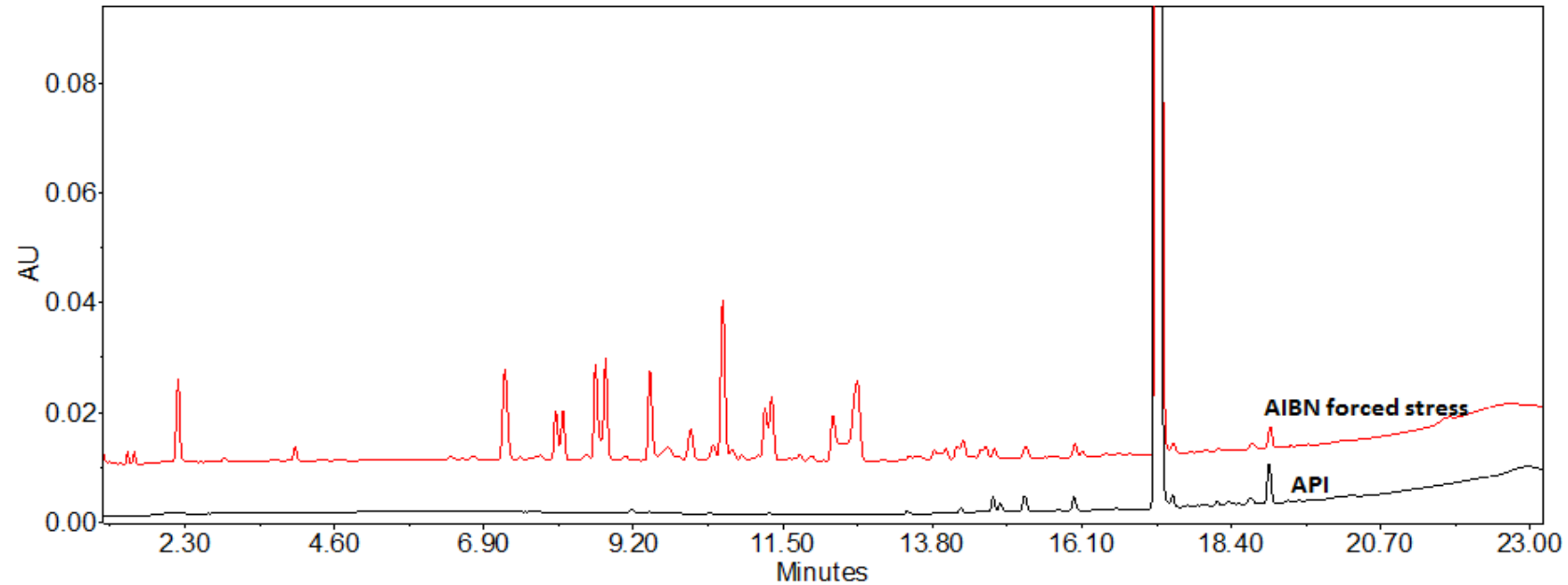
*isolation*

Deg

*NMR/MS*

Structure ID  
(Degradate)

# Typical Oxidative Forced Stress

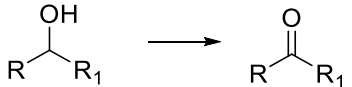
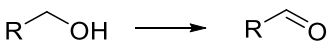
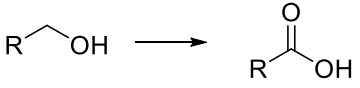
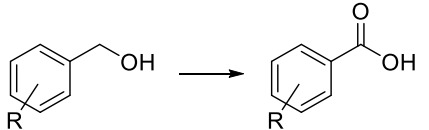


Oxidative forced stress experiment with AIBN for a MSD development candidate

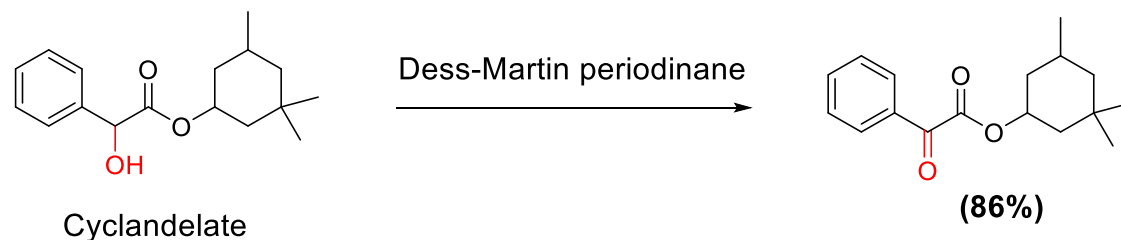
Not ideal for isolation of an oxidative degradation product!!



# Relevant Oxidative Transformations and Recommended Reagents to Enrich API

Transformation	Reagents
	i) Dess-Martin periodinane ii) Pyridinium chlorochromate iii) Pyridinium dichromate iv) RuO <sub>2</sub> .xH <sub>2</sub> O/NaIO <sub>4</sub>
	i) Dess-Martin periodinane ii) Pyridinium chlorochromate iii) Pyridinium dichromate
	i) TPAP/NMO
	i) KMnO <sub>4</sub>

# Oxidation with Dess-Martin Periodinane



**Enrichment and isolation were done in 1 day**

300 mg API in  
6 mL  $\text{CH}_2\text{Cl}_2$



Add D-M periodinane  
(700 mg)



Stir for 2h



Filter and  
concentrate

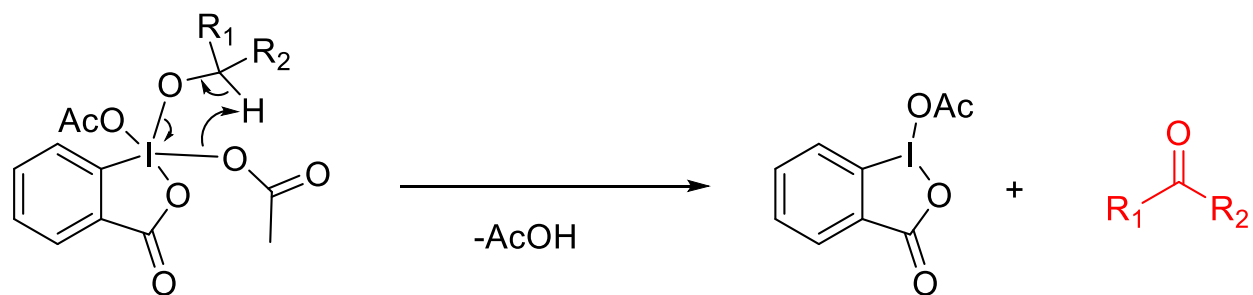
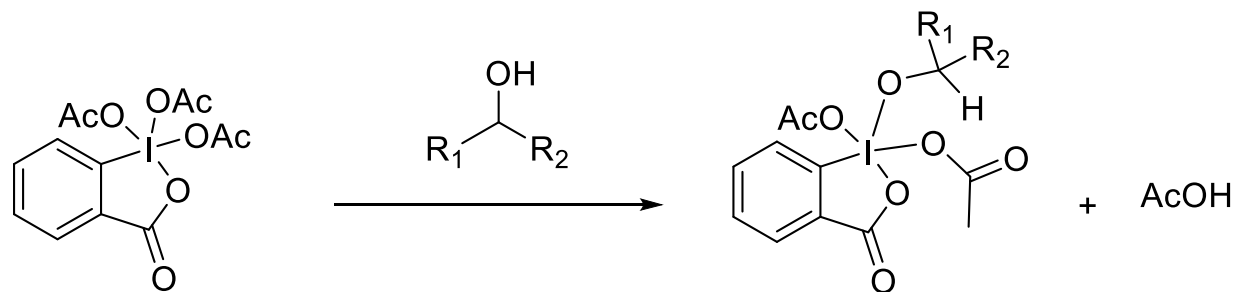


40 g silica gel  
0 to 30% ethyl acetate in hexanes  
20 minutes

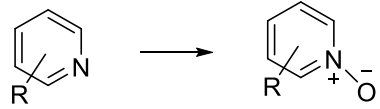
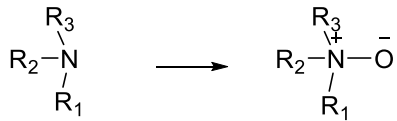
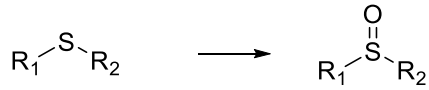
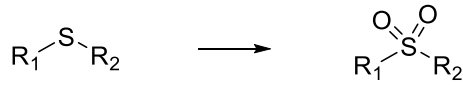
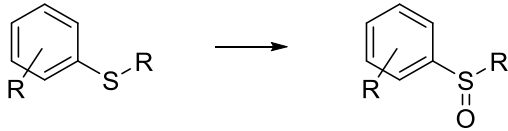
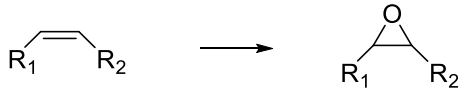


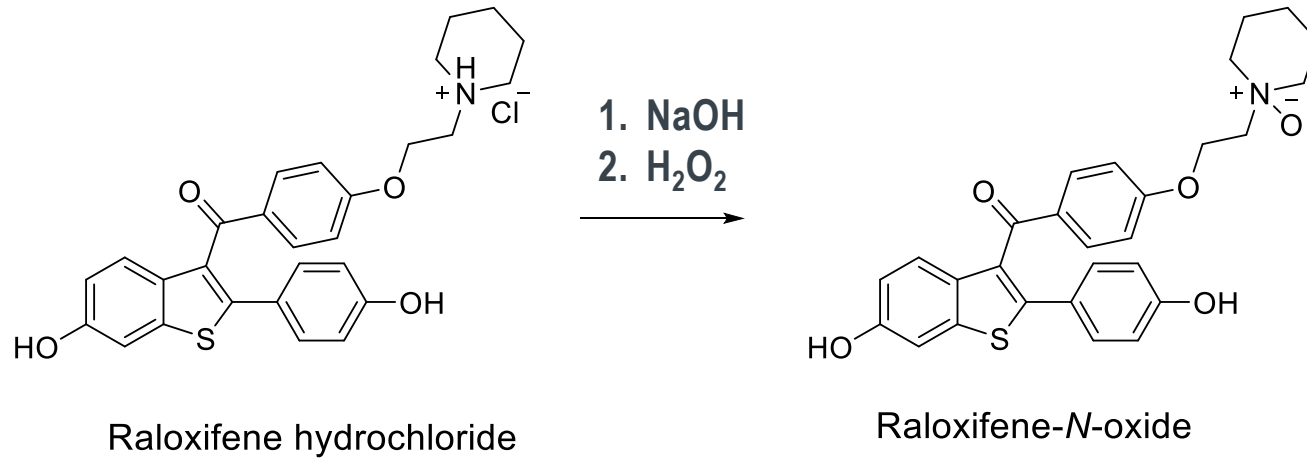
**Isolated yield: 86%**

# Mechanism: D-M Periodinane Oxidation

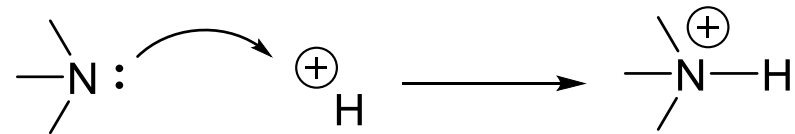
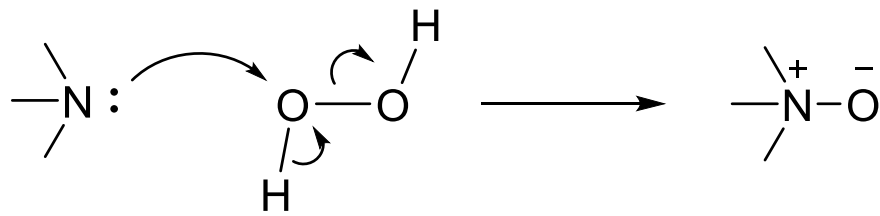


# Relevant Oxidative Transformations and Recommended Reagents to Enrich API

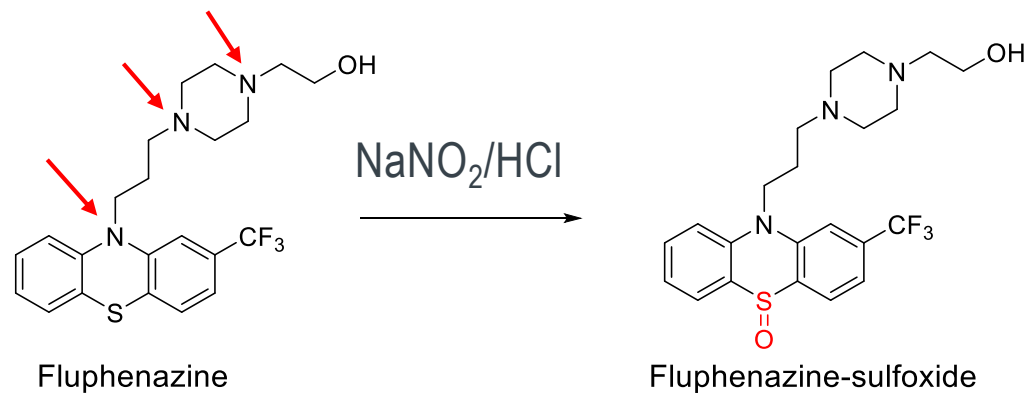
Transformation	Reagents
	i) MTO/H <sub>2</sub> O <sub>2</sub> ii) <i>m</i> -CPBA
	i) H <sub>2</sub> O <sub>2</sub> ii) Peracetic acid iii) O <sub>2</sub> /RuCl <sub>3</sub>
	i) Oxone ii) H <sub>2</sub> O <sub>2</sub> /[VO <sub>2</sub> F(dmpz) <sub>2</sub> ]
	i) Oxone
	i) NaNO <sub>2</sub> /HCl
	i) <i>m</i> -CPBA



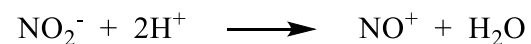
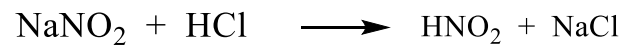
(isolated yield: 75%)



# Oxidation of Fluphenazine to Fluphenazine-Sulfoxide

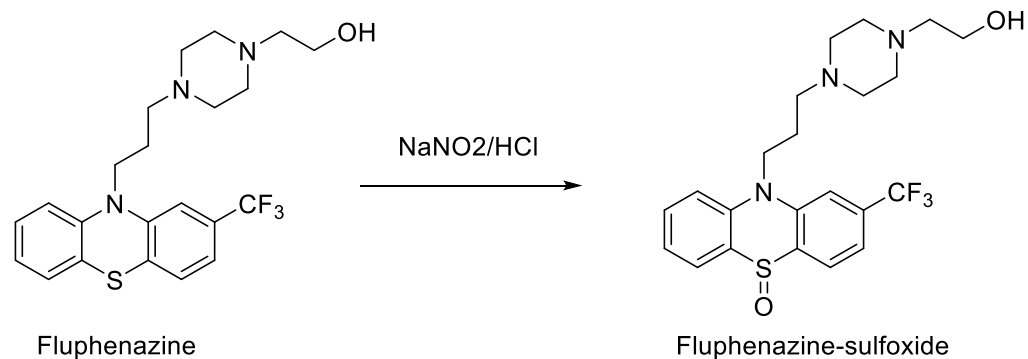


*Not a 2-electron oxidation. Oxidation via cation radical intermediate.*



↑  
Oxidant

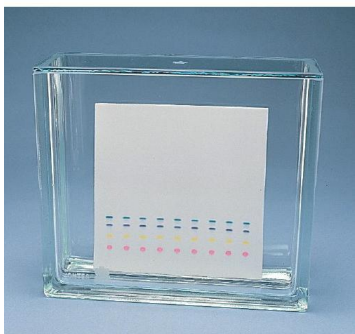
# Oxidation of Fluphenazine to Fluphenazine-Sulfoxide



22 mg API in 1 mL H<sub>2</sub>O + 2 drops conc. HCl  $\longrightarrow$  Add 2 drops aq NaNO<sub>2</sub> 100 mg/mL  $\xrightarrow{2 \text{ minutes}}$  Add NH<sub>4</sub>OH  $\longrightarrow$  Extract w CH<sub>2</sub>Cl<sub>2</sub>

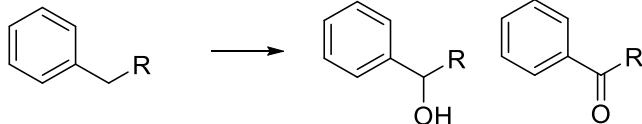
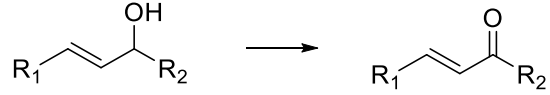
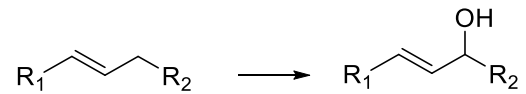
$\downarrow$

Concentrate CH<sub>2</sub>Cl<sub>2</sub> extracts



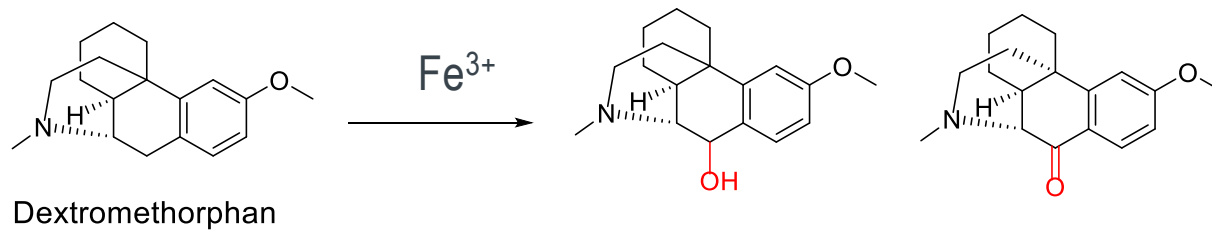
Prep-TLC  $\longrightarrow$  Isolated yield: 70%

# Relevant Oxidative Transformations and Recommended Reagents to Enrich API

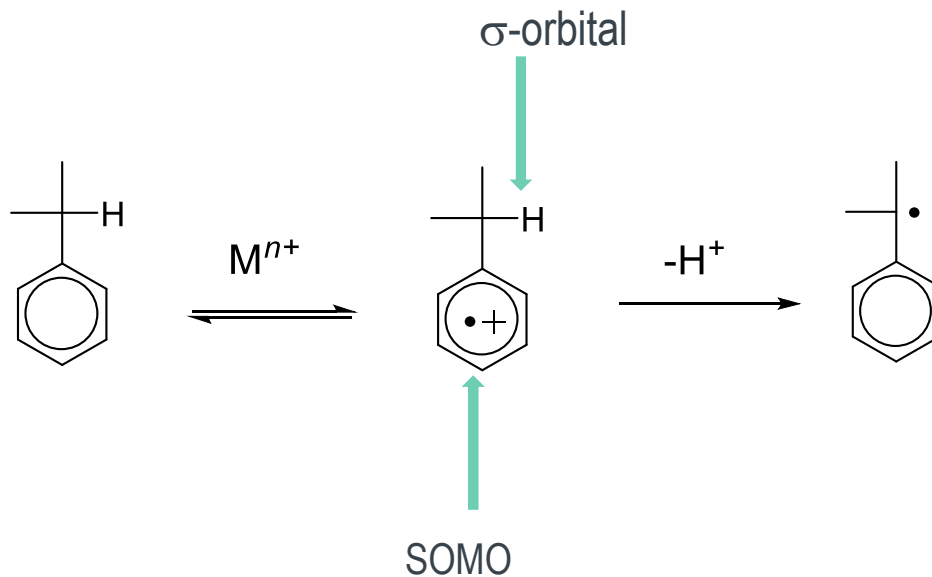
Transformation	Reagents
 <p>The diagram shows a benzene ring with a -CH<sub>2</sub>-R group. An arrow points to two products: a benzene ring with a -CH(OH)-R group and a benzene ring with a -C(=O)-R group.</p>	i) FeCl <sub>3</sub> ii) AAPH
 <p>The diagram shows a molecule with a double bond between carbons 1 and 2, and a hydroxyl group on carbon 2. The groups are labeled R<sub>1</sub> and R<sub>2</sub>. An arrow points to the product where the hydroxyl group is replaced by a carbonyl group (=O).</p>	i) H <sub>2</sub> O <sub>2</sub> (g) ii) Dess-Martin periodinane
 <p>The diagram shows a molecule with a double bond between carbons 1 and 2, and a -CH<sub>2</sub>-R<sub>2</sub> group on carbon 2. The groups are labeled R<sub>1</sub> and R<sub>2</sub>. An arrow points to the product where the -CH<sub>2</sub>-R<sub>2</sub> group is oxidized to a -CH(OH)-R<sub>2</sub> group.</p>	i) SeO <sub>2</sub>



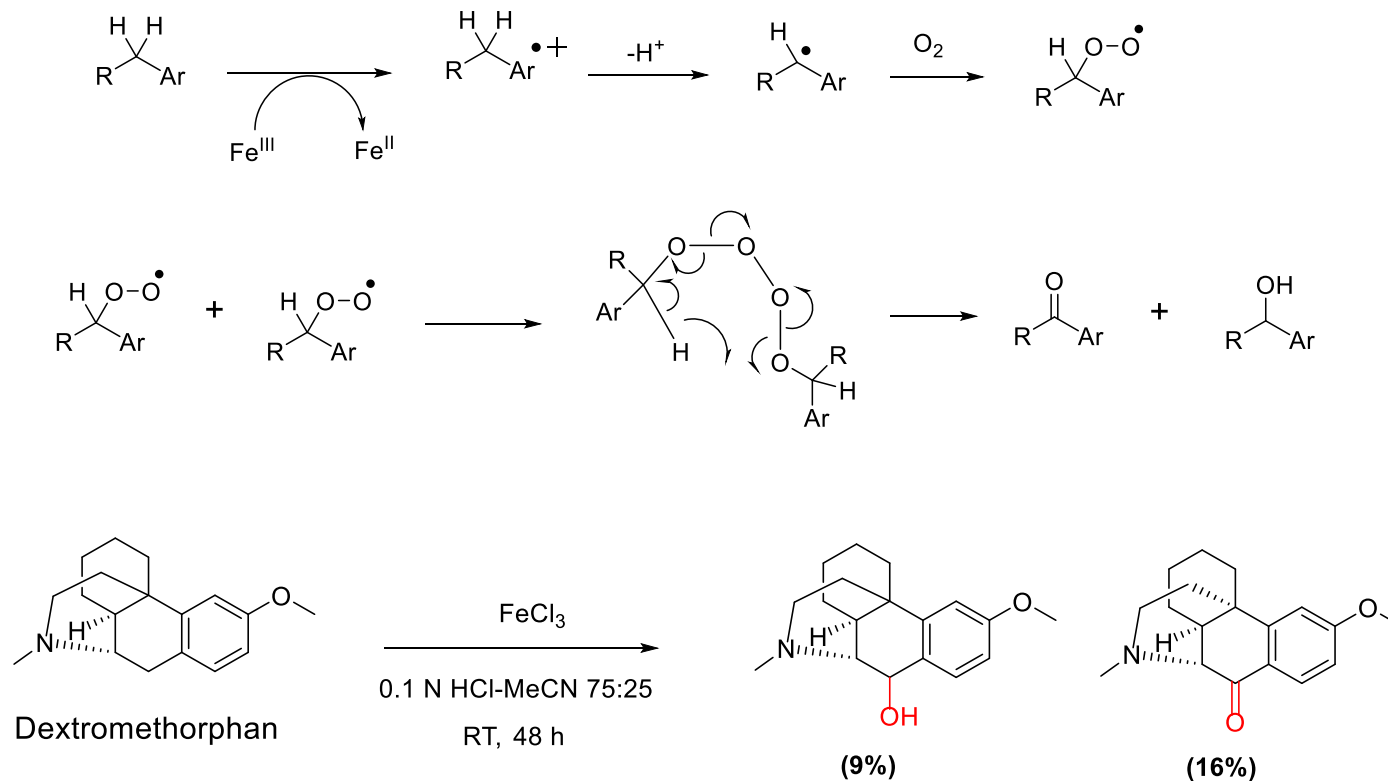
# Oxidation of Dextromethorphan



Formation of benzylic radical *via* aromatic cation radical



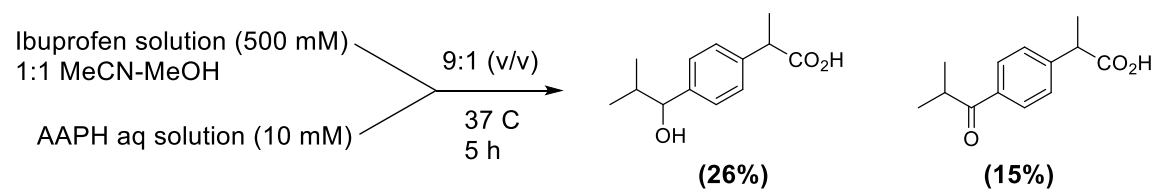
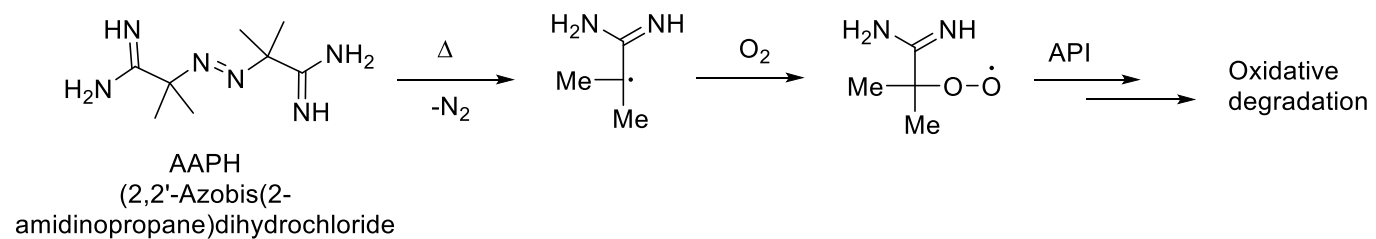
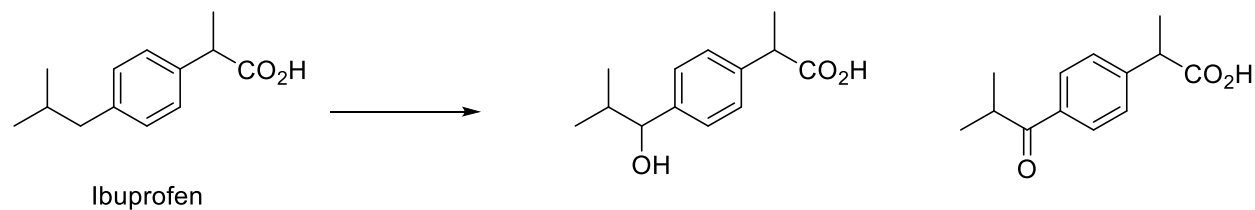
# Oxidation of Dextromethorphan



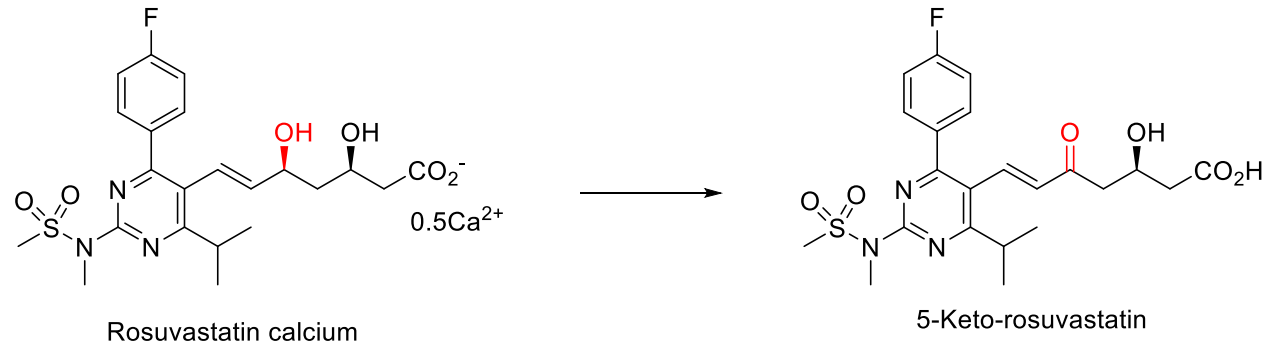
**Iron(III)-mediated oxidative degradation on the benzylic carbon of drug molecules in the absence of initiating peroxides**

Nanda, K. K.; Blincoe, W. D.; Allain, L. R.; Wuelfing, W. P.; Harmon, P. A. *J. Pharm. Sci.* 2017, 106, 1347-1354.

# Oxidation of Ibuprofen



# Oxidation of Rosuvastatin



API

H<sub>2</sub>O<sub>2</sub>

50 °C, 1 week

24% conversion to 5-keto-rosuvastatin





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Pharmaceutics, Drug Delivery and Pharmaceutical Technology

### Enrichment of Relevant Oxidative Degradation Products in Pharmaceuticals With Targeted Chemoselective Oxidation



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